

Giv	Give short answers:						
1.	. How can we change the physical state of matter?						
2.	How do liquids differ from gases?						
	How do solids differ from liquids with regard to particle arrangement?						



Class: Six Workbook: Matter as Particles Science					
4. A liquid on co	olina is converted	d into its solid state. Wha	t will		
_	olid when it is cod				
5. Write down the	e names of five lid	quids and five gases whi	ch you		
know.					
Liqu	uids	Gases			
Liqu	uids	Gases			
Liqu	uids	Gases			
Liqu	uids	Gases			
Liqu	uids	Gases			
Liqu	uids	Gases			
Liqu	uids	Gases			
Liqu	uids	Gases			
Give answers in		Gases			
Give answers in	detail:	Gases  processes in daily life:			
Give answers in 1. Explain the us	detail:	processes in daily life:			
Give answers in 1. Explain the us	detail: se of the following	processes in daily life:			



Class : Six	Workbook : Matter as Particles	Science
Boiling:		
Doming		
Evaporation:		
•		
Condensation:		
Melting:		
2. Why do solid	ds have fixed volume and fixed shape?	



. Why do solids not flow like liquids and gases?  . What is diffusion? Explain with the help of examples.	lass : Six	Workbook : Matter as Particles	Science
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. What is diffusion? Explain with the help of examples.			
	. What is diffusion	on? Explain with the help of examples.	



5. What is sublimation? Explain with the help of examples.					

Write C against correct and I against incorrect statement in the middle column. Also correct the incorrect statement and write it in the next column.

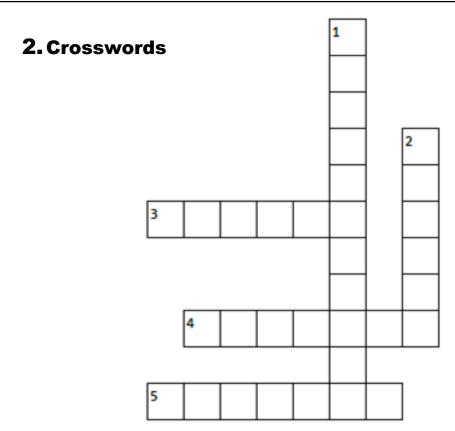
Correct / incorrect	C/I	Correct statement
Particles in solids are strongly		
packed with each other giving		
definite shape and definite		
volume to solid objects.		
Particles in a liquid have		
definite shape but no definite		
volume.		
Gases have neither definite		
shape nor definite volume.		
The distance between the		
particles of liquids are greater		
than that between the particles		
of gases.		



## 1. Multiple choice questions:

1.	There are stronge	r forces of attraction b	etwe	en the particle	es:	
a.	Solids	b. Liquids	c. G	ases	d.	All of these
2.	Solid and liquid ob	ojects cannot be comp	resse	ed easily as th	neii	particles are:
a.	Closely packed	b. Loosely packed	sp ar	acking paces mong nem		Scattered irregularly
3.	The process of ch	anging gas into liquid	:			
a.	Melting	b. Evaporation	c. Fr	reezing	d.	Condensation
4.	Changing of substistermed as:	tance directly from so	lid sta	ate to gaseous	s st	ate on heating
a.	Boiling	b. Sublimation	c. M	lelting	d.	Diffusion
5.	Material that don't	take the shape of the	cont	ainer:		
a.	Solids	b. Liquids	c. G	ases	d.	All of these
6.	When a gas conde	enses, it becomes a:				
a.	Solid	b. Liquid	c. C	rystal	d.	Another gas
7.	When a solid obje	ct is heated, its partic	les be	egin to:		
a.	Vibrate fast	b. Vibrate slowly	c. St	top brating	d.	Move freely
8.	Boiling point of wa	ater is:				
a.	0° C	b. 0° F	c. 10	00° C	d.	100° F
9.	Movement of partithey are less:	icles from an area who	ere th	ney are more t	:O 8	an area where
a.	Boiling	b. Evaporation	c. D	iffusion	d.	Sublimation
10	. Which of the	e following is opposite	to bo	oiling?		
a.	Evaporation	b. Freezing	c. M	lelting	d.	Condensation





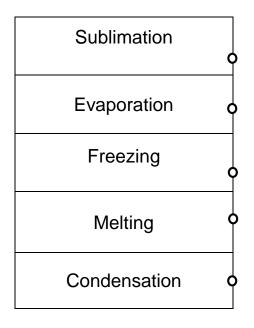
## **Across**

- 3. Made up of particles
- 4. Do not have fixed shape
- 5. Conversion of solid into liquid

#### **Down**

- 1. Conversion of liquid into gas
- 2. Particles are tightly packed together

## 3. Columns



Conversion of solid into liquid
Conversion of gas into liquid
Conversion of solid into gas
Conversion of liquid into gas
Conversion of liquid into solid



### 4. Words Search

Find the following word in the words search.

EXISTENCE	DENSITY	COMPRESSION	MOVEMENT
FORMATION	CONVERSION	POURED	SUBSTANCES

Е	Q	W	Е	R	D	Т	Υ	U	ı	0	Р	S
X	Α	S	D	F	Е	G	С	Ι	Т	J	K	U
ı	L	Z	X	С	Ν	>	0	В	Ν	Ν	М	В
S	Q	W	R	Т	S	Υ	N	U	Е	I	0	S
Т	L	K	7	Η	I	G	V	F	М	D	S	Т
Е	М	Ν	В	V	Т	O	Е	Ρ	Е	0	_	Α
N	Р	U	Υ	Т	Υ	R	R	Е	V	W	Q	Ν
С	0	М	Ρ	R	Е	S	S	_	0	Ν	Ρ	O
Е	J	L	K	J	Ι	G	ı	F	М	D	(C)	Е
Α	R	М	Z	В	V	O	0	Ρ	X	Z	Q	S
Α	Е	S	D	F	G	Ι	Ν	7	K	K	L	М
N	D	F	0	R	М	Α	Т	ı	0	N	V	С

## 5. Jumbled Words

i.	selcitrap	 ii.	rianoecnt	
iii.	rnrgaaecf	 iv.	etialceclr	
V.	inwbonra	vi.	sffniiduo	
vii.	itrdcieno	viii.	rtiomsue	
ix.	bulmatsition	 <b>x</b> .	tingmel	

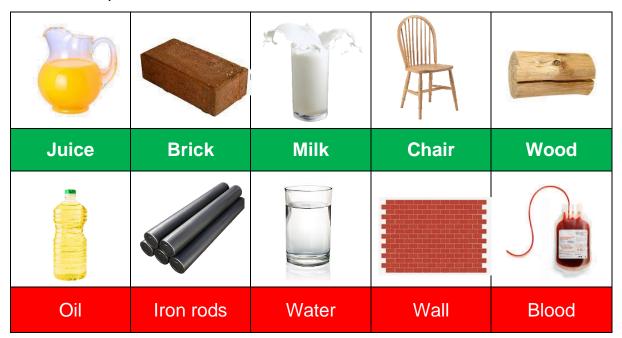
# 6. Fill in the blanks using the given words.

	Solid	condensation	Matter	diffusion	Solid		
i.	. Any substance that has mass and occupies space is called						
ii.	In		- , particles v	vibrate at their	mean		
	position.						
iii.	The state	e of matter whic	h has fixed sh	ape and volum	e is called		
iv.		substance chan		seous to a liqu	id state		
	directly,	the process is c	alled		<u>_</u> ·		
V.	Mixing o	f ink in water is	an example of		·		
7. Write "T" for the true and "F" for the false statement.							
1. Li	quid has l	esser density tha	an solids.				
2. Particles vibrate at their position in gas.							
3. Al	3. All matter is made up of small particles called atoms.						
4. Th	ne process	s of changing a	liquid into a so	lid is called fre	ezing.		
	obert brow	vn was a chemis	st who observe	ed changes in a	ì		



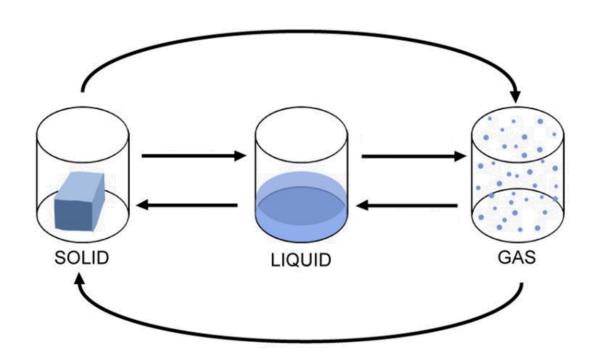
## 8. Drag and Drop

Look at the pictures and write their names in the relevant column.



Solid	Liquid

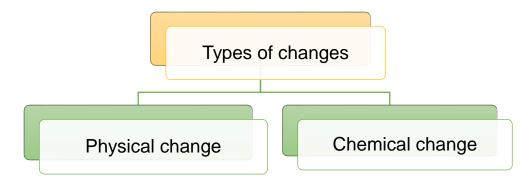
## 9. Label the diagram.



## 10. Comprehension

# Answer the following questions after reading the paragraph carefully.

Generally, matter experiences two main types of changes: chemical change and physical change. In the process of physical change, there is no alteration in the natural identity of the matter, though the change in its state, size, and shape occurs. But chemical changes are irreversible and totally different substances are formed in them. Some common changes of matter include freezing, melting, condensation, vaporization etc.





1. How many types of changes occur in matter?
2. What do you know about chemical change?
3. Write the names of some changes of matter.